



National Institute of Standards & Technology

Certificate of Analysis

Standard Reference Material[®] 2711

Montana Soil

Moderately Elevated Trace Element Concentrations

This Standard Reference Material (SRM) is intended primarily for use in the analysis of soils, sediments, or other materials of a similar matrix. SRM 2711 is a moderately contaminated soil that was oven-dried, sieved, and blended to achieve a high degree of homogeneity. A unit of SRM 2711 consists of 50 g of the dried material.

The certified elements for SRM 2711 are given in Table 1. The values are based on measurements using one definitive method or two or more independent and reliable analytical methods. Noncertified values for a number of elements are given in Table 2 as additional information on the composition. The noncertified values should **NOT** be used for calibration or quality control. Analytical methods used for the characterization of this SRM are given in Table 3 along with analysts and cooperating laboratories. All values (except for carbon) are based on measurements using a sample weight of at least 250 mg. Carbon measurements are based on 100 mg samples.

NOTICE AND WARNINGS TO USERS

Expiration of Certification: This certification of SRM 2711 is valid, within the measurement uncertainties specified, until **31 December 2011**, provided the SRM is handled in accordance with instructions given in this certificate (see Instructions for Use). This certification is nullified if the SRM is damaged, contaminated, or otherwise modified.

Maintenance of SRM Certification: NIST will monitor this SRM over the period of its certification. If substantive technical changes occur that affect the certification before the expiration of this certificate, NIST will notify the purchaser. Return of the attached registration card will facilitate notification.

Statistical consultation was provided by S.B. Schiller of the NIST Statistical Engineering Division.

The overall direction and coordination of the analyses were under the chairmanship of M.S. Epstein and R.L. Watters, Jr., of the NIST Inorganic Analytical Research Division.

The technical and support aspects involved in the preparation, certification, and issuance of this SRM were coordinated through the NIST Standard Reference Materials Program by T.E. Gills and J.S. Kane. Revision of this certificate was coordinated through the NIST Standard Reference Materials Program by B.S. MacDonald.

Willie E. May, Chief
Analytical Chemistry Division

Gaithersburg, MD 20899
Certificate Issue Date: 18 January 2002
See Certificate Revision History on Last Page

John Rumble, Jr., Acting Chief
Standard Reference Materials Program

INSTRUCTIONS FOR USE

Use: A minimum sample weight of 250 mg (dry weight - see Instructions for Drying) should be used for analytical determinations to be related to the certified values on this Certificate of Analysis.

To obtain the certified values, sample preparation procedures should be designed to effect complete dissolution. If volatile elements (i.e., Hg, As, Se) are to be determined, precautions should be taken in the dissolution of SRM 2711 to avoid volatilization losses.

Instructions for Drying: When nonvolatile elements are to be determined, samples should be dried for 2 h at 110 °C. Volatile elements (i.e., Hg, As, Se) should be determined on samples as received; separate samples should be dried as previously described, to obtain a correction factor for moisture. Correction for moisture is to be made to the data for volatile elements before comparing to the certified values. This procedure ensures that these elements are not lost during drying. The approximate weight loss on drying has been found to be in the range of 1.5 % to 2.2 %.

Source and Preparation of Material: The U.S. Geological Survey (USGS), under contract to the NIST, collected and processed the material for SRM 2711. The material is an agricultural soil collected in the till layer (upper 15.2 cm (6 in)) of a wheat field. The soil from a 3.05 m × 3.05 m (10 ft × 10 ft) area was shoveled into 0.114 m³ (3 gal) plastic pails for shipment to the USGS laboratory for processing.

The material was spread on 30.5 cm × 61 cm (1 ft × 2 ft) polyethylene-lined drying trays in an air drying oven and dried for three days at room temperature. The material was then passed over a vibrating 2 mm screen to remove plant material, rocks, and large chunks of aggregated soil. Material remaining on the screen was deaggregated and rescreened. The combined material passing the screen was ground in a ball mill to pass a 74 μm screen and blended for 24 h. Twenty grab samples were taken and measured for the major oxides using X-ray fluorescence spectrometry and for several trace elements by using inductively coupled plasma atomic emission analysis to provide preliminary assessment of the homogeneity prior to bottling. The material was bottled into 50 g units and randomly selected bottles were taken for the final homogeneity testing.

Analysis: The homogeneity, using selected elements in the bottled material as indicators, was assessed using X-ray fluorescence spectrometry and neutron activation analysis. In a few cases, statistically significant differences were observed, and the variance due to material inhomogeneity is included in the overall uncertainty of the certified values. The estimated relative standard deviation is less than 3 % for those elements for which homogeneity was assessed.

Certified Values and Uncertainties: The certified values are weighted means of results from two or more analytical methods, or the mean of results from a single definitive method, except for mercury. Mercury certification is based on cold vapor atomic absorption spectrometry used by two different laboratories employing different methods of sample preparation prior to measurement. The weights for the weighted means were computed according to the iterative procedures of Paule and Mandel [1]. The stated uncertainty includes allowances for measurement imprecision, material variability, and differences among analytical methods. Each uncertainty is the sum of the half-width of a 95 % prediction interval and includes an allowance for systematic error among the methods used. In the absence of systematic error, a 95 % prediction interval predicts where the true concentrations of 95 % of the samples of this SRM lie. The certified values were corroborated by analyses from nine Polish laboratories cooperating on the certification under the direction of T. Plebanski and J. Lipinski, Polish Committee for Standardization Measures, and Quality Control. The Polish laboratory work was supported by the Maria Sklodowska-Curie Joint Fund.

Table 1. Certified Values

Element	Mass Fraction (%)	Element	Mass Fraction (µg/g)
Aluminum	6.53 ± 0.09	Antimony	19.4 ± 1.8
Calcium	2.88 ± 0.08	Arsenic	105 ± 8
Iron	2.89 ± 0.06	Barium	726 ± 38
Magnesium	1.05 ± 0.03	Cadmium	41.70 ± 0.25
Phosphorus	0.086 ± 0.007	Copper	114 ± 2
Potassium	2.45 ± 0.08	Lead	1162 ± 31
Silicon	30.44 ± 0.19	Manganese	638 ± 28
Sodium	1.14 ± 0.03	Mercury	6.25 ± 0.19
Sulfur	0.042 ± 0.001	Nickel	20.6 ± 1.1
Titanium	0.306 ± 0.023	Selenium	1.52 ± 0.14
		Silver	4.63 ± 0.39
		Strontium	245.3 ± 0.7
		Thallium	2.47 ± 0.15
		Vanadium	81.6 ± 2.9
		Zinc	350.4 ± 4.8

Noncertified Values: Noncertified values, shown in parentheses, are provided for information only. An element concentration value may not be certified, if a bias is suspected in one or more of the methods used for certification, or if two independent methods are not available.

Table 2. Noncertified Values

Element	Mass Fraction (%)	Element	Mass Fraction (µg/g)
Carbon	(2)	Bromine	(5)
		Cerium	(69)
		Cesium	(6.1)
		Chromium	(47)
		Cobalt	(10)
		Dysprosium	(5.6)
		Europium	(1.1)
		Gallium	(15)
		Gold	(.03)
		Hafnium	(7.3)
		Holmium	(1)
		Indium	(1.1)
		Iodine	(3)
		Lanthanum	(40)
		Molybdenum	(1.6)
		Neodymium	(31)
		Rubidium	(110)
		Samarium	(5.9)
		Scandium	(9)
		Thorium	(14)
		Tungsten	(3)
		Uranium	(2.6)
		Ytterbium	(2.7)
		Yttrium	(25)
		Zirconium	(230)

Table 3. Analytical Methods Used for the Analysis of SRM 2711

Element	Certification Methods *	Element	Certification Methods *
Ag	ID ICPMS; RNAA; INAA	Mo	ID ICPMS
Al	XRF1; XRF2; INAA; DCP; ICP	Na	INAA; FAES
As	RNAA; HYD AAS; INAA	Nd	ICP
Au	INAA; FAAS	Ni	ID ICPMS; ETAAS; INAA
Ba	XRF2; FAES; ICP; INAA	P	DCP; COLOR; XRF2; ICP
Br	INAA	Pb	ID TIMS; POLAR; ICP
C	COUL	Rb	INAA
Ca	XRF1; XRF2; DCP; INAA; ICP	S	ID TIMS
Cd	ID ICPMS; RNAA	Sb	INAA; ETAAS
Ce	INAA; ICP	Sc	INAA; ICP
Co	INAA; ETAAS; ICP	Se	RNAA; HYD AAS; INAA
Cr	INAA; DCP; ICP	Si	XRF1; XRF2; GRAV
Cs	INAA	Sm	INAA
Cu	RNAA; FAES; ICP	Sr	ID TIMS; INAA; ICP
Dy	INAA	Th	ID TIMS; INAA; ICP
Eu	INAA	Ti	INAA; XRF1; XRF2; DCP
Fe	XRF1; XRF2; DCP; INAA	Tl	ID TIMS; LEAFS
Ga	INAA; ICP	U	ID TIMS
Hf	INAA	V	INAA; ICP
Hg	CVAAS	W	INAA
Ho	INAA	Y	ICP
I	INAA	Yb	INAA; ICP
In	INAA	Zn	ID TIMS; ICP; INAA; POLAR
K	XRF1; XRF2; FAES; ICP; INAA	Zr	INAA
La	INAA; ICP		
Mg	XRF1; ICP		
Mn	INAA; ICP; XRF2; XRF1		

*Methods in **bold** were used to corroborate certification methods or to provide information values.

COLOR	Colorimetry; lithium metaborate fusion.
COUL	Combustion coulometry.
CVAAS	Cold vapor atomic absorption spectrometry.
DCP	Direct current plasma atomic emission spectrometry; lithium metaborate fusion.
ETAAS	Electrothermal atomic absorption spectrometry; mixed acid digestion.
FAAS	Flame atomic absorption spectrometry; mixed acid digestion, except for Au, leached with HBr-Br ₂ .
FAES	Flame atomic emission spectrometry; mixed acid digestion.
GRAV	Gravimetry; sodium carbonate fusion.
HYD AAS	Hydride generation atomic absorption spectrometry.
ICP	Inductively coupled plasma atomic emission spectrometry; mixed acid digestion.
ID ICPMS	Isotope dilution inductively coupled plasma mass spectrometry; mixed acid digestion.
ID TIMS	Isotope dilution thermal ionization mass spectrometry; mixed acid digestion.
INAA	Instrumental neutron activation analysis.
LEAFS	Laser enhanced atomic fluorescence spectrometry; mixed acid digestion.
POLAR	Polarography.
RNAA	Radiochemical neutron activation analysis; mixed acid digestion.
XRF1	Wavelength dispersive X-ray fluorescence on fused borate discs.
XRF2	Wavelength dispersive X-ray fluorescence spectrometry on pressed powder.

Participating NIST Analysts:

M. Adriaens	A. Marlow
E.S. Beary	J.R. Moody
C.A. Beck	P.J. Paulsen
D.S. Braverman	P. Pella
M.S. Epstein	T.A. Rush
J.D. Fassett	J.M. Smeller
K.M. Garrity	G.C. Turk
R.R. Greenberg	T.W. Vetter
W.R. Kelly	R.D. Vocke
R.M. Lindstrom	L.J. Wood
E.A. Mackey	R.L. Watters, Jr.

Participating Laboratories:

P. Briggs, D. Siems, J. Taggart, S. Wilson U.S. Geological Survey Branch of Geochemistry Denver, CO, USA	P. Bienkowski Institute of Ecology Dziekanow Lesny, Poland
J.B. Bodkin College of Earth and Mineral Sciences The Pennsylvania State University University Park, PA, USA	H. Matusiewicz Technical University Poznan, Poland
S.E. Landsberger, V.G. Vermette Department of Nuclear Engineering University of Illinois Urbana, IL, USA	B. Ksiazek Geological Enterprise Warsaw, Poland
J. Lipinski, T. Plebanski Polish Committee for Standardization, Measures and Quality Control Warsaw, Poland	G. Szoltyk Forest Research Institute Division in Sekocin, Warsaw, Poland
M. Bielawska, B. Galczynska, J. Galczynska, K. Galczynski, K. Wiacek Institute of Soil Science and Plant Cultivation Pulawy, Poland	J. Rojek District Chemical Agricultural Station Bydgoszcz, Poland
I. Matuszczyk Forest Research Institute Division in Katowice, Warsaw, Poland	E. Gorecka Polish Geological Institute Warsaw, Poland
Z. Jonca Institute of Environmental Protection Warsaw, Poland	

REFERENCE

- [1] Paule, R.C. and Mandel, J., *NBS Journal of Research* 87, pp. 377-385, (1982).

Certificate Revision History: 18 January 2002 (This revision reflects a change in the certification expiration date); 23 August 1993 (Addendum added); 30 October 1992 (Original certificate date).

Users of this SRM should ensure that the certificate in their possession is current. This can be accomplished by contacting the SRM Program at: telephone (301) 975-6776; fax (301) 926-4751; e-mail srminfo@nist.gov; or via the Internet <http://www.nist.gov/srm>.

Addendum to SRM Certificates
2709 San Joaquin Soil
2710 Montana Soil
2711 Montana Soil

Leachable Concentrations Using U.S. EPA Method 3050 for Flame Atomic
Absorption Spectrometry (FAAS) and Inductively Coupled Plasma Atomic Emission
Spectrometry (ICP-AES)

The certified concentrations of constituent elements in essentially all National Institute of Standards and Technology (NIST) chemical composition Standard Reference Materials (SRMs) are given as total concentrations. The certified concentrations are based on measurements obtained by two or more independent methods or techniques. The measurement methods require complete sample decomposition, or the sample may be analyzed nondestructively. Where complete sample decomposition is required, it can be accomplished by digestion with mixed acids or by fusion. For mixed acid decomposition, hydrofluoric acid must be included in the acid mixture used to totally decompose siliceous materials, such as soils and sediments.

For a number of environmental monitoring purposes, the concentrations of labile or extractable fractions of elements are more useful than total concentrations. Concentrations of labile or extractable fractions are generally determined using relatively mild leach conditions, which are unlikely to totally decompose the sample. It should be noted that results obtained using the mild leach conditions are often erroneously depicted in reports as total concentrations. However, reported concentrations of labile or extractable fractions of elements are generally lower than total concentrations; recovery can be total if an element in a given sample is completely labile. Results are often presented as measured concentration in the leachate in comparison to the total or certified concentration. The recovery of an element as a percent of total concentration is a function of several factors such as the mode of occurrence in the sample, leach medium, leach time and temperature conditions, and pH of the sample-leach medium mixture. References [1] through [27] may be consulted for detailed discussions of these factors and their effect on leach results. Some of these references provide leach data for one or more reference materials.

In its monitoring programs, the U.S. Environmental Protection Agency (EPA) has established a number of leach methods for the determination of labile or extractable elements. They include Methods 3015, 3050, and 3051. A number of cooperating laboratories using the variation to U.S. EPA Method 3050 for Flame Atomic Absorption Spectrometry (FAAS) and Inductively Coupled Plasma-Atomic Emission Spectrometry (ICP-AES) measurements, have reported data for SRMs 2709, 2710, and 2711. This variation of the method uses hydrochloric acid in its final step, which is different from Method 3050 for ICP-MS and Hydride Generation-Atomic Absorption Spectrometry (HG-AAS) measurements. The data obtained are presented in Tables 1, 2, and 3 of this addendum. The names of the cooperating laboratories are listed in Table 4. Several laboratories provided replicate (3 to 6) analyses for each of the three soil SRMs. The number of results for a given element varied from only one to as many as nine, as indicated in the data presented in Tables 1 through 3. Because of the wide range of interlaboratory results for most elements, only the data range and median of the individual laboratory means are given. Ranges differ somewhat from those in Reference [26], since this addendum is based on a larger data set than had been available previously.

For SRMs 2710 and 2711, 17 laboratories provided data as part of contract work for the U.S. EPA. Each SRM was treated as a blind sample in one quarter of 1992. Since there was no within-laboratory replication of analysis in the design of the exercise, the 17-laboratory means of results were treated as single laboratory results from laboratories using replication, in establishing the median of the full data set. In a few cases, however, the contract laboratories mean was the only result available for a particular element (e.g., Antimony in SRM 2710). In others, the contract laboratories mean is also the median for the full leach data set (e.g., Arsenic in SRM 2710). An asterisk identifies those cases where the contract laboratories means are given as the median value.

Please note none of the values in Tables 1 through 3 are certified, but are given as information on the performance of the three soils when used to evaluate, or to provide quality control for Method 3050 followed by FAAS and ICP-AES measurements only. The data should not be used for any other purpose. **The certified values, provided as total concentrations, are the best estimate of the true concentrations.**

Table 1. Leach Data from Cooperating Laboratories for Soil SRM 2709

Element	Range		Median	N	% Leach Recovery†
	Wt %				
Aluminum	2.0	- 3.1	2.6	5	35
Calcium	1.4	- 1.7	1.5	5	79
Iron	2.5	- 3.3	3.0	8	86
Magnesium	1.2	- 1.5	1.4	5	93
Phosphorus	0.05	- 0.07	0.07	3	100
Potassium	0.26	- 0.37	0.32	5	16
Silicon	---	---	< 0.01	1	< 1
Sodium	0.063	- 0.11	0.068	4	6
Titanium	0.03	- 0.04	0.038	3	11
	mg/kg				
Antimony	---	---	< 10	1	...
Arsenic	---	---	< 20	2	...
Barium	392	- 400	398	2	41
Cadmium	---	---	< 1	5	...
Chromium	60	- 115	79	5	61
Cobalt	10	- 15	12	5	90
Copper	26	- 40	32	7	92
Lead	12	- 18	13	5	69
Manganese	360	- 600	470	7	87
Molybdenum	---	---	< 2	2	...
Nickel	65	- 90	78	7	89
Selenium	nr	- nr	0.014	1	< 1
Strontium	100	- 112	101	3	44
Vanadium	51	- 70	62	3	55
Zinc	87	- 120	100	7	94

$$\dagger \% \text{ Leach Recovery} = 100 \times \left[\frac{\text{Median Value}}{\text{Certified/Information Value}} \right]$$

--- at or below the detection limit

... no % Leach Recovery calculated

nr no range reported by the laboratory

Table 2. Leach Data from Cooperating Laboratories for Soil SRM 2710

Element	Range		Median	N	% Leach Recovery†	
	Wt %					
Aluminum	1.2	-	2.6	1.8	6	28
Calcium	0.38	-	0.48	0.41	7	33
Iron	2.2	-	3.2	2.7	9	80
Magnesium	0.43	-	0.60	0.57	6	67
Phosphorus	0.106	-	0.11	0.11	2	100
Potassium	0.37	-	0.50	0.45	6	21
Silicon	---	-	---	< 0.01	1	< 1
Sodium	0.049	-	0.062	0.054	5	5
Titanium	0.092	-	0.11	0.10	3	35
	mg/kg					
Antimony	3.4	-	12	7.9*	1*	21
Arsenic	490	-	600	590	3	94
Barium	300	-	400	360	3	51
Cadmium	13	-	26	20	8	92
Chromium	15	-	23	19	6	(49)
Cobalt	6.3	-	12	8.2	7	(82)
Copper	2400	-	3400	2700	8	92
Lead	4300	-	7000	5100	8	92
Manganese	6200	-	9000	7700	8	76
Mercury	27	-	37	32*	1*	98
Molybdenum	13	-	27	20	2	(100)
Nickel	8.8	-	15	10.1	8	71
Silver	24	-	30	28	3	79
Selenium	nr	-	nr	0.002	1	...
Strontium	94	-	110	100	3	(42)
Thallium	0.50	-	0.76	0.63*	1*	(48)
Vanadium	37	-	50	43	4	56
Zinc	5200	-	6900	5900	9	85

$$\dagger \% \text{ Leach Recovery} = 100 \times \left[\frac{\text{Median Value}}{\text{Certified/Information Value}} \right]$$

() indicates that information value was used

--- at or below the detection limit

... no % Leach Recovery could be calculated

nr no range reported by the laboratory

* U.S. EPA contact laboratories mean; treated as one laboratory since no within-laboratory replication; see text

Table 3. Leach Data from Cooperating Laboratories for Soil SRM 2711

Element	Range		Median	N	% Leach Recovery†	
Wt %						
Aluminum	1.2	-	2.3	1.8	5	28
Calcium	2.0	-	2.5	2.1	5	73
Iron	1.7	-	2.6	2.2	7	76
Magnesium	0.72	-	0.89	0.81	5	77
Phosphorus	0.06	-	0.09	0.088	3	100
Potassium	0.26	-	0.53	0.38	5	16
Silicon	---		---	< 0.01	1	< 1
Sodium	0.020	-	0.029	0.026	4	2.3
Titanium	0.039	-	0.048	0.042	2	14
mg/kg						
Antimony	< 10	1	...
Arsenic	88	-	110	90	3	86
Barium	170	-	260	200	2	28
Cadmium	32	-	46	40	6	96
Chromium	15	-	25	20	4	(43)
Cobalt	7	-	12	8.2	5	(82)
Copper	91	-	110	100	6	88
Lead	930	-	1500	1100	7	95
Manganese	400	-	620	490*	7	77
Molybdenum	---		---	< 2	2	...
Nickel	14	-	20	16	7	78
Silver	2.5	-	5.5	4.0	1	86
Selenium	nr	-	nr	0.009	1	< 1
Strontium	48	-	55	50	3	20
Vanadium	34	-	50	42	3	51
Zinc	290	-	340	310	7	89

$$\dagger \% \text{ Leach Recovery} = 100 \times \left[\frac{\text{Median Value}}{\text{Certified/Information Value}} \right]$$

() indicates that information value was used

--- at or below the detection limit

... no % Leach Recovery could be calculated

nr no range reported by the laboratory

* U.S. EPA contact laboratories mean; treated as one laboratory since no within-laboratory replication; see text

Table 4. Leach Study for Cooperating Laboratories

SRMs 2709, 2710, and 2711

S.A. Wilson; U.S. Geological Survey, Lakewood, CO, USA

J. Lipinski and T. Plebanski; Polish Committee for Standardization, Measures and Quality Control, Warsaw, Poland

E. Gorecka; Polish Geological Institute, Warsaw, Poland

M. Paul; Research Institute of Vegetable Crops, Skierniewice, Poland

I. Matuszczyk; Forest Research Institute, Warsaw, Poland

Z. Jonca; Institute of Environmental Protection, Warsaw, Poland

B. Ksiazek; Geological Enterprise, Warsaw, Poland

I. Twardowska; Polish Academy of Sciences, Institute of Environmental Engineering, Zabrze, Poland

SRMs 2710 and 2711

L. Butler and D. Hillman; U.S. Environmental Protection Agency, Las Vegas, NV, and 17 contract laboratories

REFERENCES

- [1] Gallagher, P.H. and Walsh, T., *Proceedings of the Royal Irish Academy*, **49B**, pp. 1-17, (1943).
- [2] McKeague, J.A. and Day, J.A., *Canadian Journal of Soil Science*, **46**, pp. 13-22, (1966).
- [3] Arshad, M.A., St. Arnaud, R.J., and Huang, P.M., *Canadian Journal of Soil Science*, **52**, pp. 19-26, (1972).
- [4] Chao, T.T. and Zhou, L., *Soil Science Society of America Journal*, **47**, pp. 225-232, (1983).
- [5] Olade, M.E. and Fletcher, K., *Journal of Geochemical Exploration*, **3**, pp. 337-344, (1974).
- [6] Chao, T.T., *Journal of Geochemical Exploration*, **20**, pp. 101-135, (1984).
- [7] Lepp, N.W., ed., *Effect of Heavy Metal Pollution on Plants, Vol.1 Effects of Trace Metals on Plant Function*, Applied Science Publishers, London, (1981).
- [8] Sondag, F., *Journal of Geochemical Exploration*, **15**, pp. 645-652, (1981).
- [9] Federal Register 40 CFR Part 136.
- [10] Binstock, D.A., Grohse, P.M., Gaskill, A. Jr., Sellers, C., Kingston, H.M., and Jassie, L.B., *Journal of Association of Official Analytical Chemists*, **74(#2)**, pp. 360-366, (1991). (Data: SRMs 2704, 4355, 1086, 1634b on 2704)
- [11] Kingston, H.M. and Walter, P.J., *Spectroscopy*, **7**, pp. 20-27, (1992).
- [12] Martens, D.C., Chesters, G., and Peterson, L.A., *Soil Science Society of American Proceedings*, **29**, pp. 411-413, (1966).
- [13] Trefry, J.H. and Metz, S., *Analytical Chemistry*, **56**, pp. 745-749, (1984).
- [14] Rendell, P.S., Batley, G.E., and Cameron, J.A., *Environmental Science and Technology*, **14**, pp. 314-318, (1980).
- [15] Wilson, S.A., Unpublished Report to NIST Standard Reference Materials Program, (1992). (Data: SRMs 2709, 2710, 2711, and 2704).
- [16] Lipinski, J., Unpublished Report to NIST Standard Reference Materials Program, (1992). (Data: SRMs 2709, 2710, 2711, and 2704).
- [17] Butler, L. and Hillman, D., Unpublished Report to NIST Standard Reference Materials Program, Quarterly Blind Summary from CLP, (1992). (Data: SRM 2710, 17 laboratories).
- [18] Kanm-iin, W.R., and Brandt, M.J., *Spectroscopy*, **4(#3)**, pp. 49-52, (1989). (Data: EPA solid QC sample).
- [19] Harper, S.L., Walling, J.F., Holland, D.M., and Pranger, L.J., *Analytical Chemistry*, **55(#9)**, pp. 1553-1557, (1983). (Data: SRM 1648).
- [20] Hewitt, A.D. and Reynolds, C.M., *Atomic Spectroscopy*, **11(#5)**, pp. 187-192, (1990). (Data: SRM 2704, RMA Soil Standard).
- [21] Nieuwenhuijze, J., Poley-Vos, C.H., van den Akker, A.H., and van Delft, W., *Analyst*, **116**, pp. 347-351, (1991). (Data: Jafa and BCR CRMs).
- [22] Hewitt, A.D. and Cragin, J.H., *Environmental Science and Technology*, **25**, pp. 985-986, (1991).
- [23] Hewitt, A.D. and Cragin, J.H., *Environmental Science and Technology*, **26**, p. 1848, (1992).
- [24] Hewitt, A.D. and Cragin, J.H., U.S. Army Toxic and Hazardous Materials Agency Report CETHATS-CR-92061, (1992).
- [25] Hewitt, A.D. and Reynolds, C.M., U.S. Army Toxic and Hazardous Materials Agency Report CETHATS-CR-90052, (1990).
- [26] Rasberry, S.D. and Kane, J.S., *American Environmental Laboratory*, **2/93**, pp. 34-35, (1993). (Data: SRMs 2709, 2710, 2711; a subset of the data reported in this addendum).
- [27] Kane, J.S., Wilson, S.D., Lipinski, J. and Butler, L., *American Environmental Laboratory*, **6/93**, pp. 14-15, (1993).