



National Institute of Standards & Technology

Certificate of Analysis

Standard Reference Material[®] 3127a

Spectrometric Standard Solution

Lanthanum

Batch Code 691806

This Standard Reference Material (SRM) is intended primarily for use in calibrating instruments used in atomic spectrometry, including atomic absorption spectrometry, inductively coupled plasma optical spectrometry, and inductively coupled plasma mass spectrometry. It can also be used in conjunction with any other analytical technique or procedure where an aqueous standard solution is required. One unit of SRM 3127a consists of 50 mL of a single element solution prepared gravimetrically from high purity lanthanum oxide (La_2O_3) to contain a known amount of lanthanum in an approximate nitric acid volume fraction of 10 %.

The certified value (Y) is based on replicate titrations against a reference solution of high purity lanthanum metal. Elemental impurity levels were determined by optical emission spectrometry. The value has been adjusted upward by 0.1 % relative, based on estimated transpiration losses of solvent through the container walls of 0.2 % relative per year. The density of the solution is $1.073 \text{ g/mL} \pm 0.002 \text{ g/mL}$ at $22 \text{ }^\circ\text{C}$.

Certified Value (Y) of Lanthanum..... $9.940 \text{ mg/mL} \pm 0.012 \text{ mg/mL}$ at $22 \text{ }^\circ\text{C}$

The uncertainty in the certified value is calculated as

$$U = (2u_c + 0.001Y) \text{ mg/mL}$$

where u_c is the "combined uncertainty" calculated according to the ISO Guide [1]. The value u_c is intended to represent, at the level of one standard deviation, the combined effect of uncertainty components associated with volumetric and gravimetric factors, as well as the purity of the starting material. The additional quantity, $0.001Y$, is an allowance for transpiration of the solution through the container walls, which is estimated to be $\pm 0.1 \%$ of the certified value during the one-year period of validity of the certification.

The combined uncertainty consists of a Type A component associated with replicate titrations and Type B components due to uncertainty in the material purity, material handling, and dilution.

This SRM was prepared by T.A. Butler of the NIST Analytical Chemistry Division. Titrimetric analyses were performed by J. Smeller and T.W. Vetter of the NIST Analytical Chemistry Division.

The support aspects involved in the preparation, certification, and issuance of this SRM were coordinated through the Standard Reference Materials Program by B.S. MacDonald.

Gaithersburg, MD 20899
Certificate Issue Date: July 30, 1996

Thomas E. Gills, Chief
Standard Reference Materials Program

Procedures for Use

Stability: This certification is valid for one year from the shipping date, provided the solution is kept tightly capped and stored under normal laboratory conditions. NIST will monitor the stability of representative solutions from this SRM lot, and if any changes occur that invalidate this certification, NIST will notify purchasers.

Preparation of Working Standard Solutions: All solutions should be brought to $22\text{ }^{\circ}\text{C} \pm 1\text{ }^{\circ}\text{C}$ before use and all glass or plastic surfaces coming in contact with the standard must have been previously cleaned. A working standard solution can be prepared from the SRM solution by serial dilution. Each dilution should be acidified with an appropriate high purity acid in high purity water. The analyst should prepare daily working solutions from 100 $\mu\text{g}/\text{mL}$ dilutions of the original SRM solution.

REFERENCE

- [1] *Guide to the Expression of Uncertainty in Measurement*, ISBN 92-67-10188-9, 1st Ed. ISO, Geneva, Switzerland, (1993); see also Taylor, B.N. and Kuyatt, C.E., "Guidelines for Evaluating and Expressing the Uncertainty of NIST Measurement Results," NIST Technical Note 1297, (1994).